

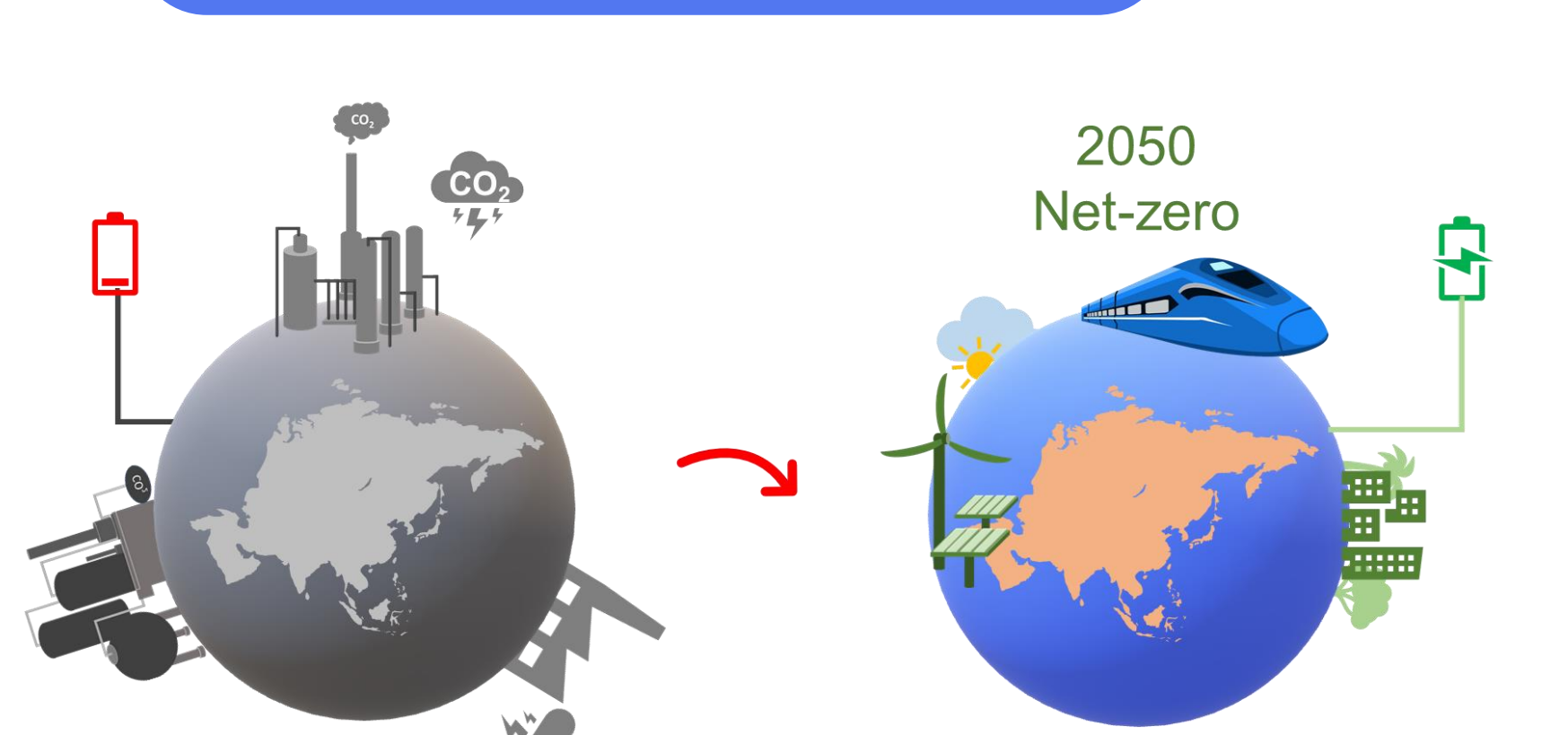
Efficient alkaline water electrolysis at high current densities via designing the metals-doped nickel sulfide nanorod arrays

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Introduction

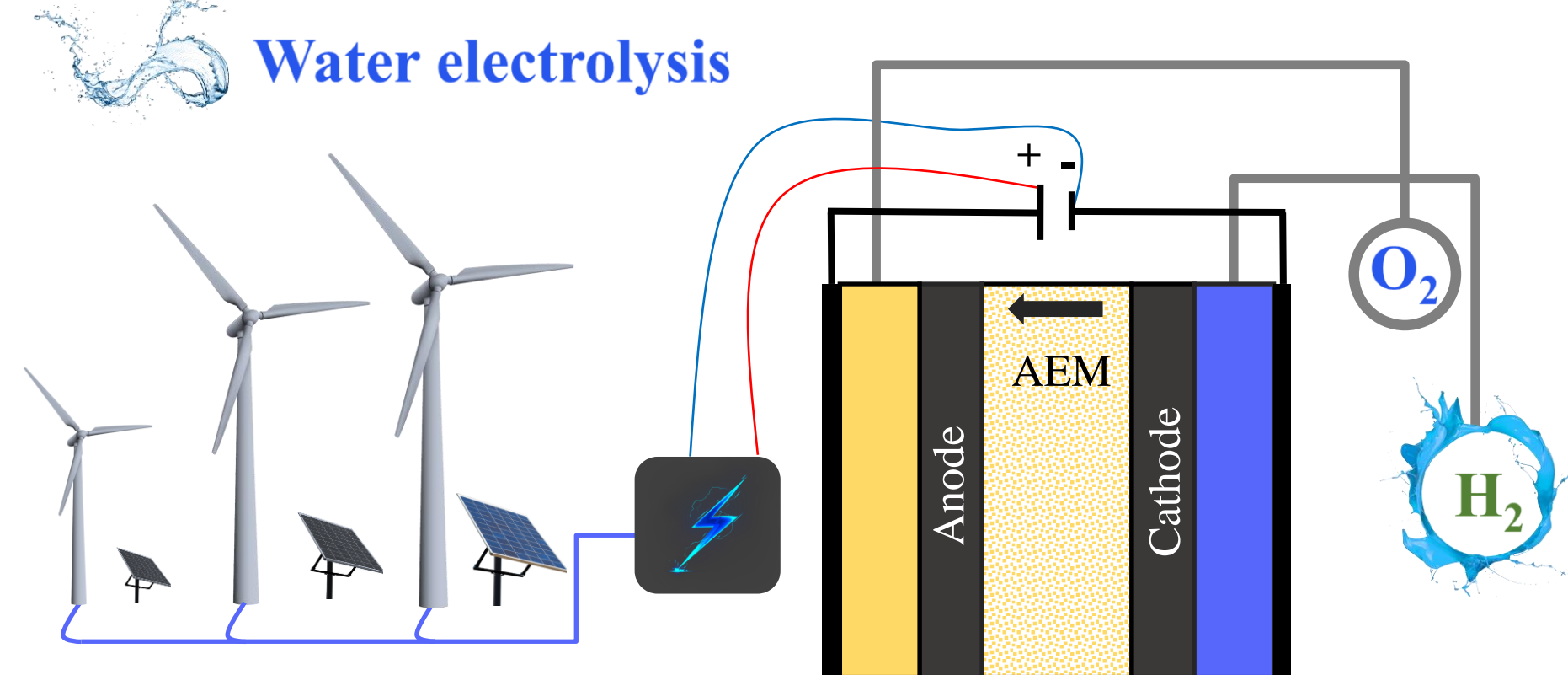
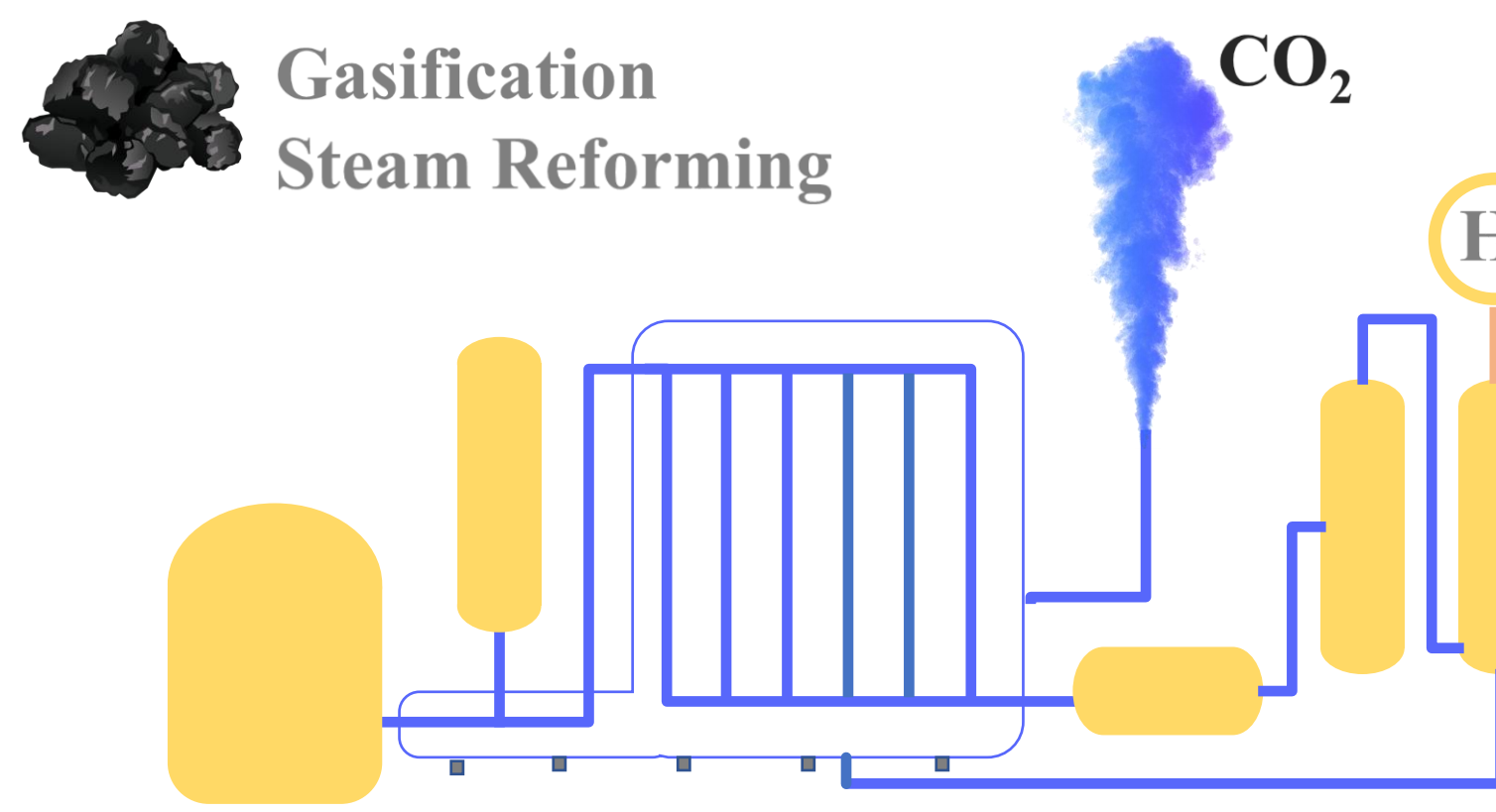


Water Splitting Green Hydrogen Production: Toward Net-Zero and Greener Future

Cathode: Hydrogen Evolution Reaction (HER)



Anode: Oxygen Evolution Reaction (OER)



Research Highlights

Bifunctional **Li, V co-doped Ni₃S₂** catalysts were prepared by a hydrothermal method.

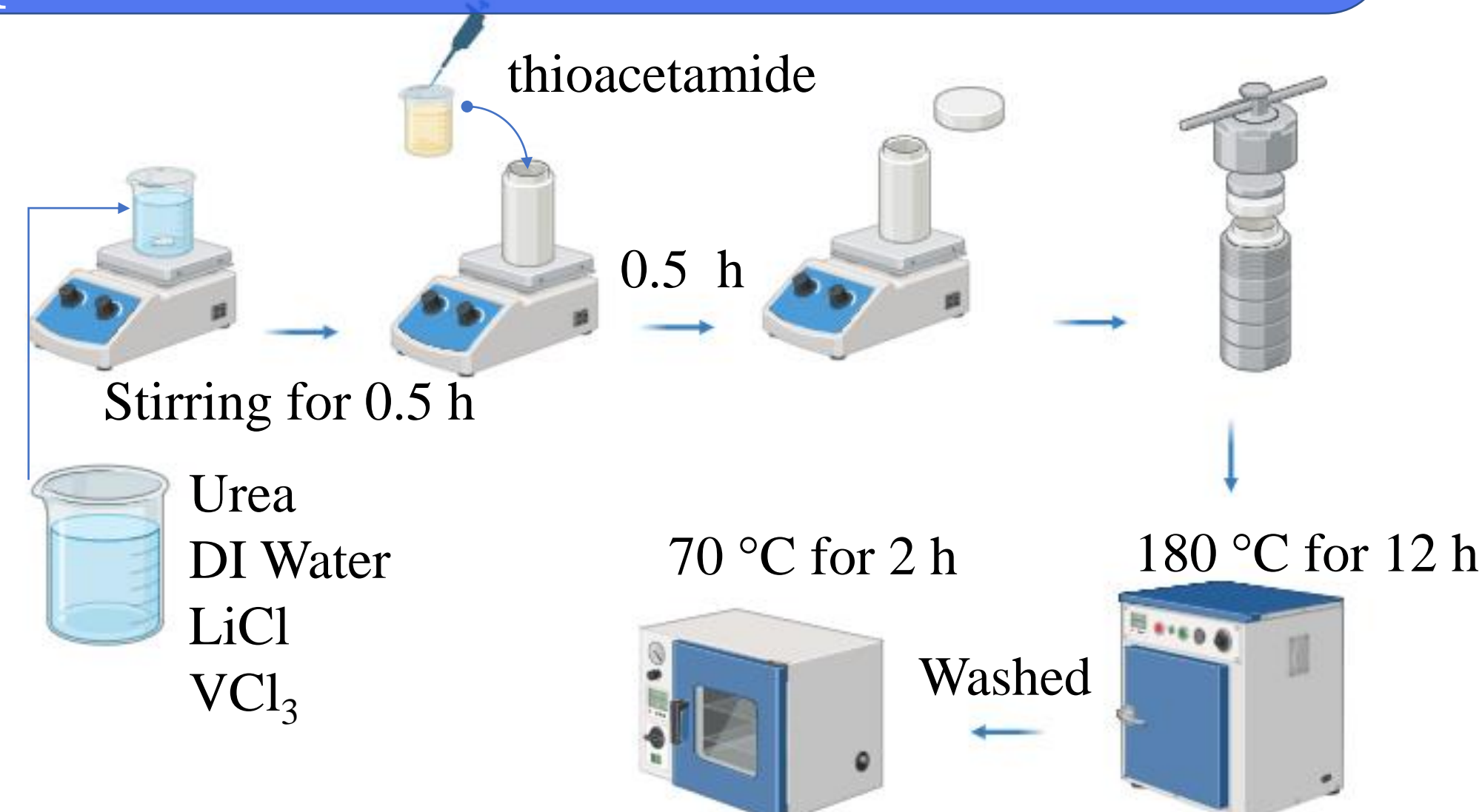
Adding Li and V into the Ni₃S₂ enhanced **intrinsic activity** and **reaction kinetics**.

Required low overpotentials of **298** and **-432 mV** for **OER** and **HER**, respectively, at a current density of **1000 mA/cm²**

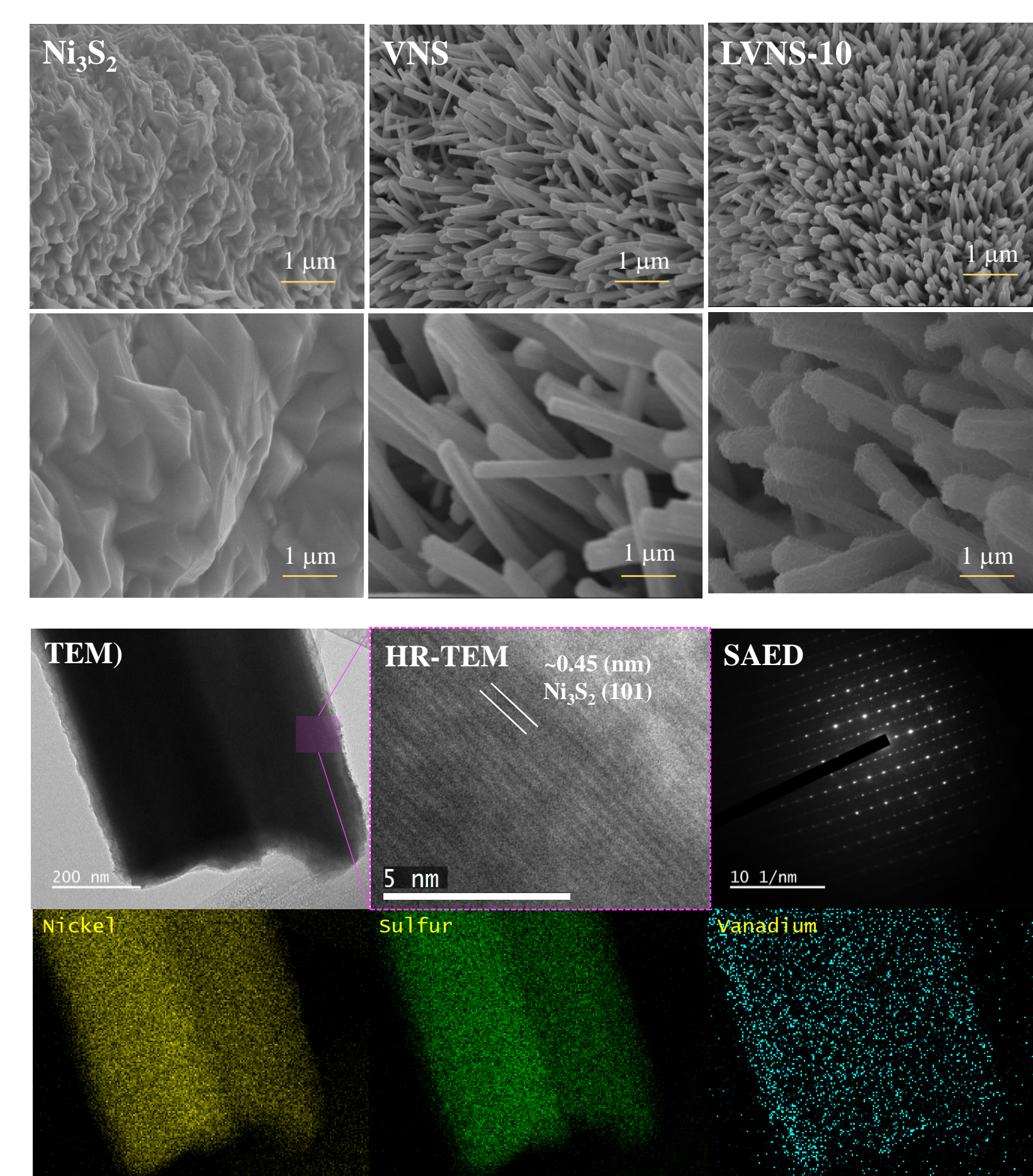
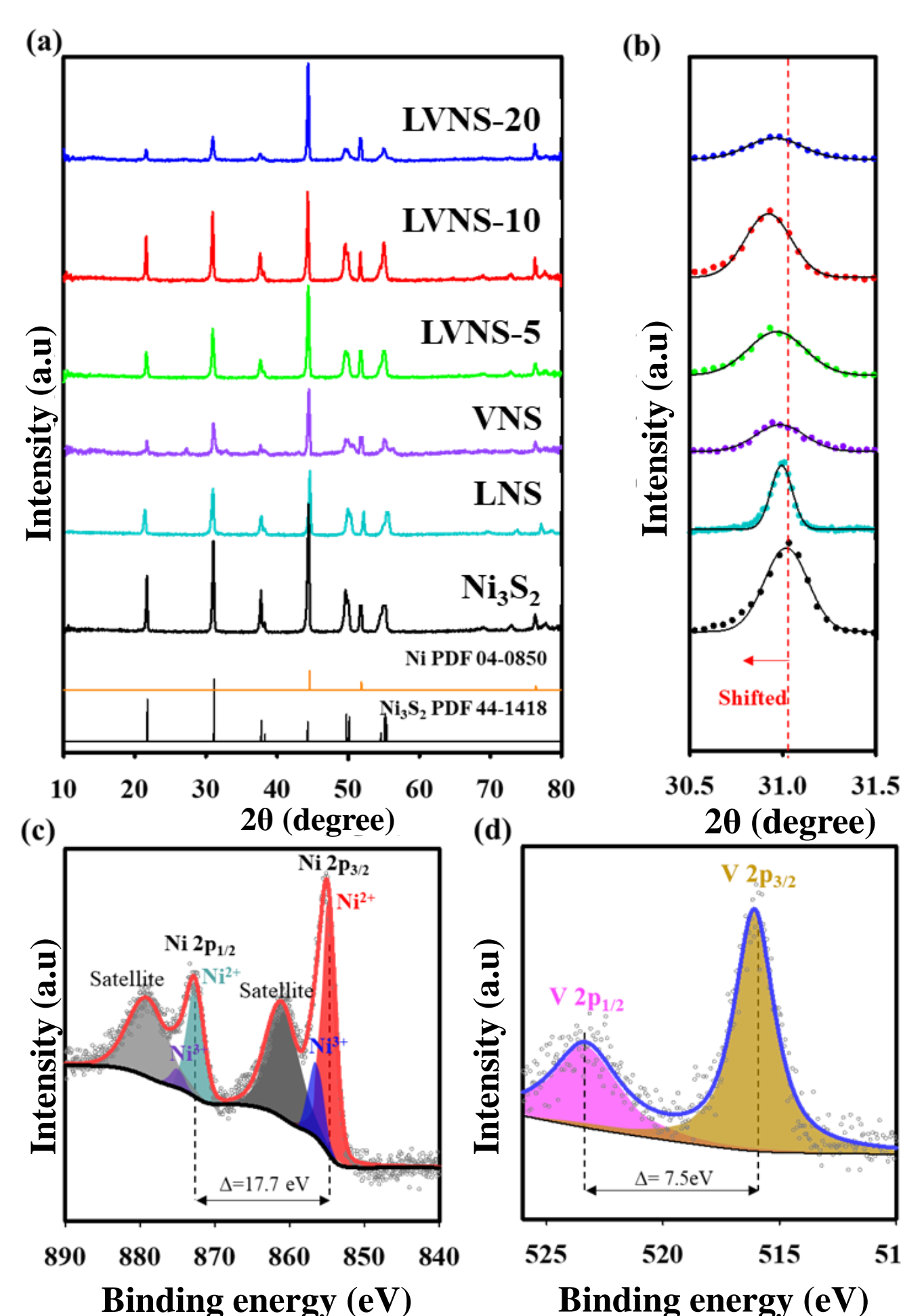
Single LVNS-10 stack cell of 2 × 2 cm² needed **1.92 V@500 mA/cm²** and **2.02 V@1000 mA/cm²**.

Our stack cell tested at 1000 mA/cm² or 4000 mA for 200 h exhibited **no degradation**.

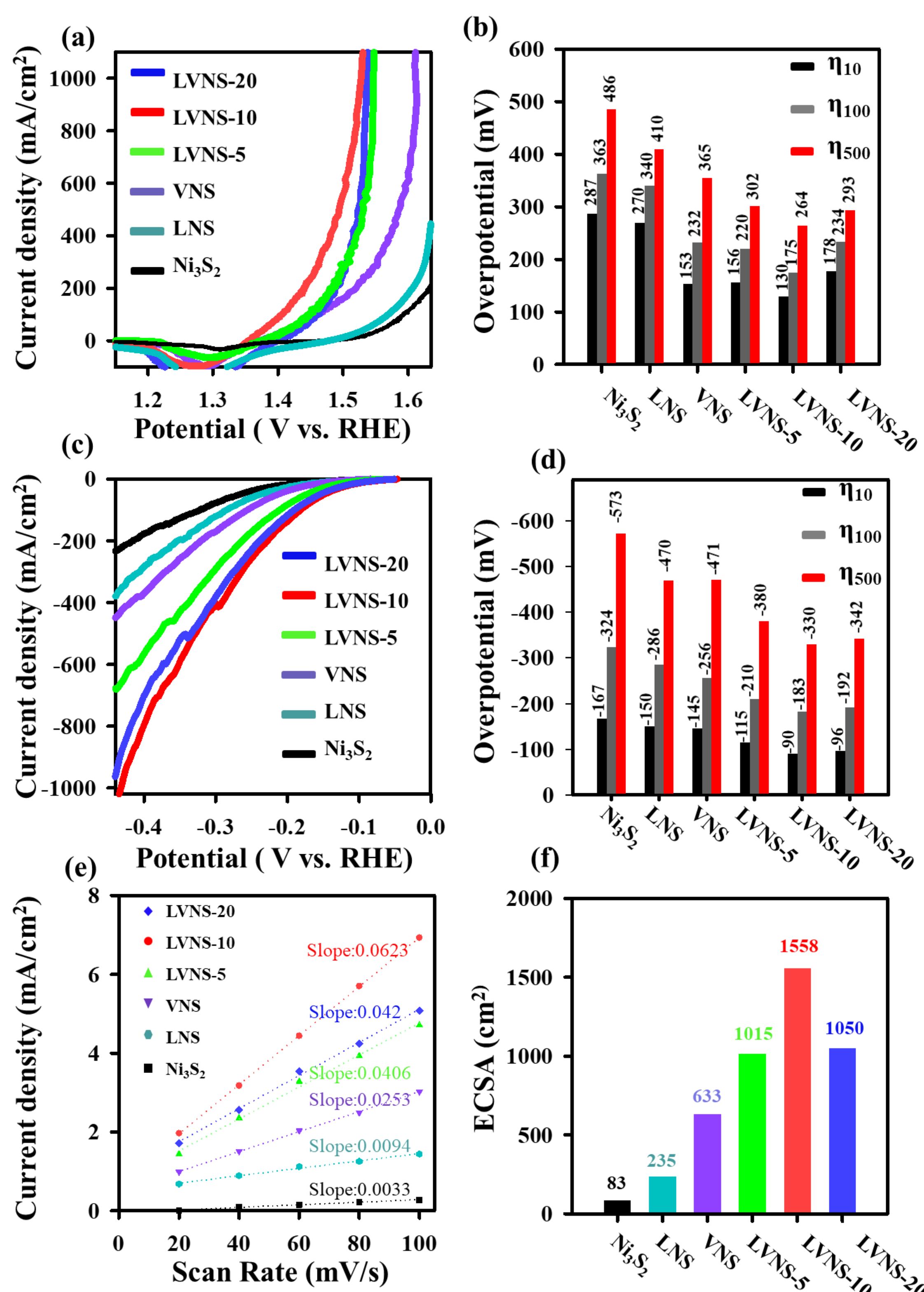
Experimental & Characterization



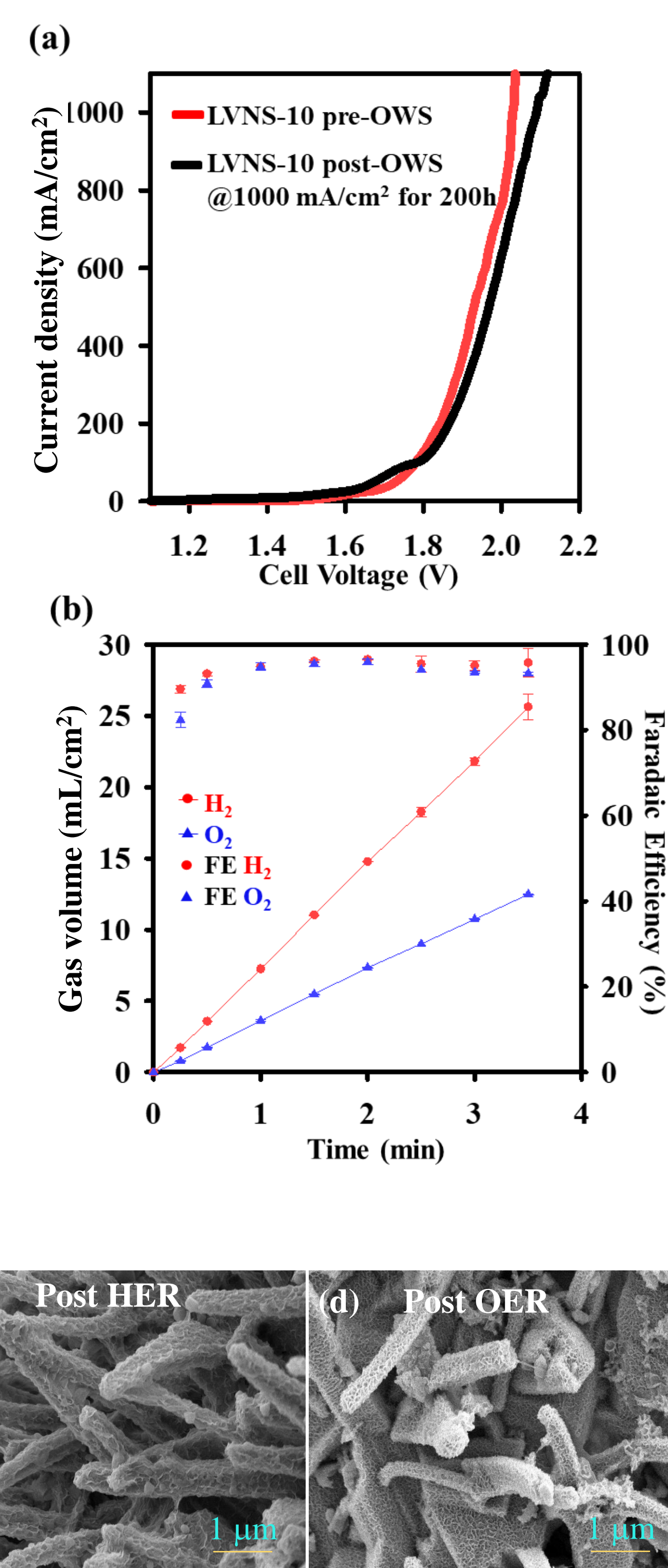
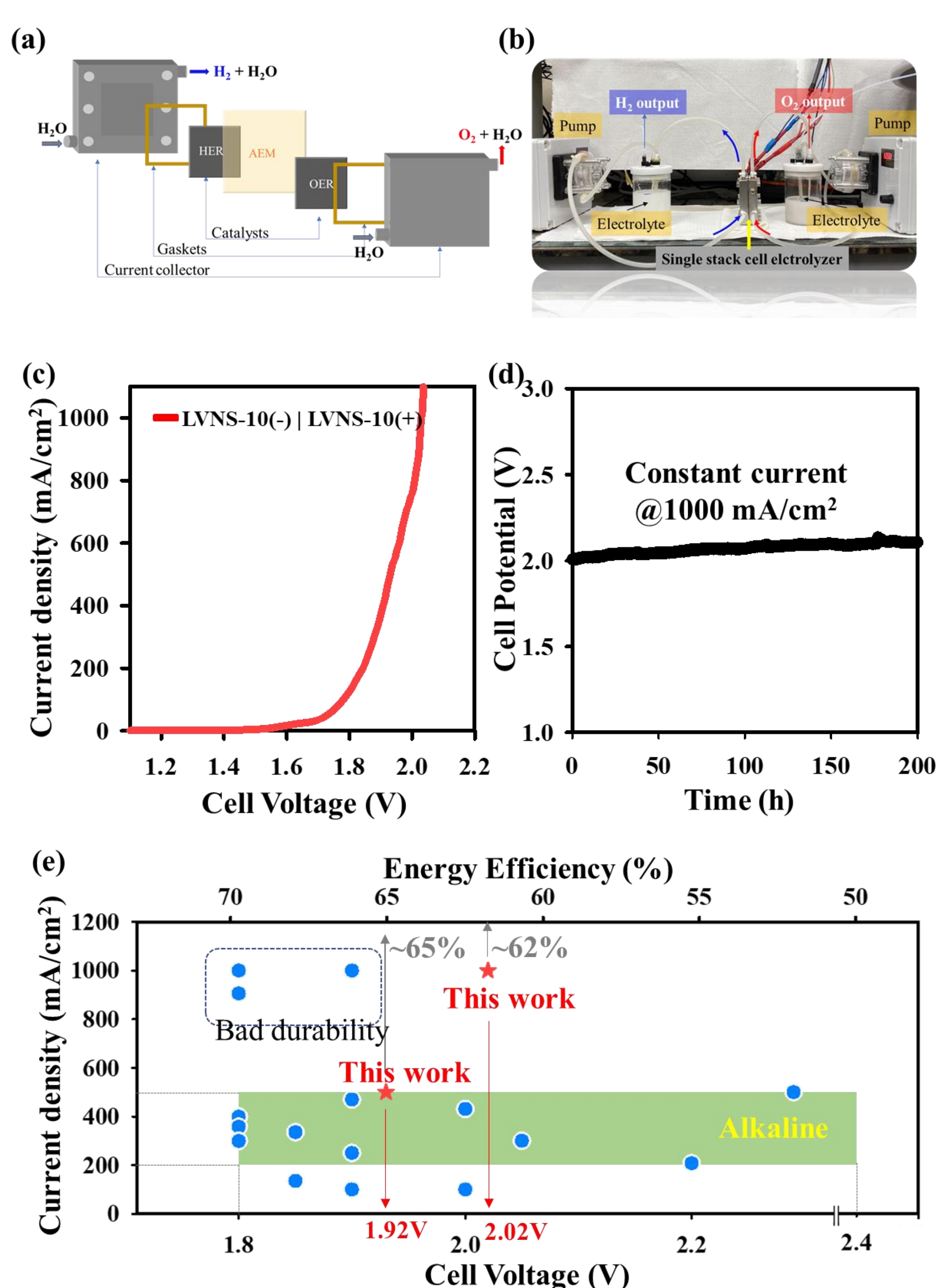
Sample name		Molar ratio			
		LiCl	VCl ₃	Urea	TAA
Pristine Ni ₃ S ₂	Ni ₃ S ₂	0	0	5	5
Li doped Ni ₃ S ₂	LNS	1	0	5	5
V doped Ni ₃ S ₂	VNS	0	0.1	5	5
V,Li-Ni ₃ S ₂	LVNS-5	1	0.05	5	5
	LVNS-10	1	0.1	5	5
	LVNS-20	1	0.2	5	5



Electrolysis Performance



LVNS-10(+)||LVNS-10(+) electrolyzer (2 × 2 cm²) performance



Conclusion

We have developed a robust bifunctional electrocatalyst based on Ni₃S₂ with ultralow potentials even when operating at excessively high current densities to make water splitting more economically competitive for potential industrial applications. The introduction of Li, V could potentially modify surface properties of Ni₃S₂, leading to enhanced catalytic activity. With the optimal Li/V ratio of 1:0.1, our LVNS-10 catalyst exhibited excellent catalytic activity for overall water splitting, regardless of low or large current densities, with remarkable durability in an alkaline electrolyte.